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fatty acid (b) $C_{17}H_{35}COOH + NaOH \rightarrow C_{17}H_{35}COO-Na^+ + H_2O$ 1
ALLOW $C_{17}H_{35}COONa$

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halide with ammonia you'll get an amine. but if you have a lot of ammonia sloshing about it will react with the HCl produced to make some NH_4Cl . This is a good example of how "real life" chemistry can be slightly different from "textbook" chemistry.

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